STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI-86 (For candidates admitted during the academic year 2008–09)

SUBJECT CODE: CH/ME/EC34

B.Sc. DEGREE EXAMINATION, NOVEMBER 2009 BRANCH IV- CHEMISTRY THIRD SEMESTER

COLIDO	E MAJOREI	••••••					
PAPER	E: MAJOR EL : ENVIRONM : 30 MINUTE	MENTAL CHEMIS	STRY	MAX.MARKS: 30			
		(30x1=30)					
		ESTION PAPER I	TSELF:				
	all the questions	e: ver the following:					
CHOOSE	the correct ans w	er the following.					
1.	The lower region of the atmosphere which extends upto 100 Km is						
2.	a) Homosphere b) Heterosphere c) Exosphere d) mesosphere gas combines rapidly with hemoglobin and reduces blood						
	capacity for carrying oxygen.						
	a) CO_2	b) <i>CO</i>	c) SO_2	d) NO_2			
3.	The contamination of water by causes Typhoid Fev a) Salmonella b) Vibrio Cholerae c) Shigella d) Entar						
	a) Salmonella	b) Vibrio Chol	erae c) Shigella	a d) Entamoeba			
4.			y pollutant released f				
	a) Cr		c) smoke				
5.			damage to the central				
	a) Hg	b) Cr	c) As	d) <i>Cd</i>			
Fill in th	ne blank:						
6.	is the antidote for arsenic poisoning.						
7.	The common trace metal present in superphophate fertilizers is						
8.	Water samples withBOD value are considered to						
	polluted.						
9.	By anaerobic fermentation in paddy fields gas is emitted.						
10.	Dissolved oxygen can be removed by using						
State wh	nether the follow	ing statements are	true or false.				
	_						
11.	Terpene hydrocarbons are emitted in significant quantities by plants and animals.						
12.	In exosphere, ozone formation takes place.						
13.							
14.	Wilson's diseases is caused by the excessive concentration of copper in the						
15.	tissues. Fluorosis causes hypermineralisation of skeleton and painful joints in human						
13.	beings.	es hypermineransau	on or skereton and pa	annai joints ni numan			

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Match the following:

16. Solar radiation reflected by clouds – sort formation.

17. Anthracite – Albedo.

18. Linear alkyl Sulfonates – Carcinogen.

19. Asbestos – Sodium formaldehyde sulfoxide.

20. *Hg* poisoning – biodegradable.

Answer in a line or two:

21. What is stratopause?

- 22. Define Lapse rate.
- 23. What is the origin of NO_x pollution?
- 24. Define aerosol with an example.
- 25. What are the additives that are used as brighteners and solublisers?
- 26. What is BOD? Mention its significance.
- 27. Name any two carcinogens released from dyeing industries along with their carcinogenic effect.
- 28. What is a mutagenic substance?
- 29. What are the main constituents of Smog?
- 30. Define L.D.₅₀ value.

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COURSE: MAJOR ELECTIVE

PAPER : ENVIRONMENTAL CHEMISTRY

TIME : 2½ HOURS MAX.MARKS : 70

SECTION - B 5X6=30

Answer any five questions.

- 1. Discuss the composition of soil.
- 2. How is the green house effect produced? Mention any four sources of Green house gases.
- Write short notes on sources and characteristics of effluents from tannery and detergents industries.
- 4. Explain indoor air pollution with examples.
- 5. What is eutrophication? Explain the types and control of eutrophication.
- 6. What are the measures to be taken to control the emission of NO_x and SO_x .
- 7. Write a note on Bhopal gas tragedy.

SECTION - C 2X20=40

Answer any two questions.

- 8. Briefly explain the segments of the environment.
- 9. Write short notes on the following:
 - a) acid rain and its adverse effects.
 - b) CFC's consequences and alternatives.

(10+10)

- 10. a) Discuss the various steps involved in the treatment of municipal and domestic waste water.
 - b) How is the Fluoride and dissolved oxygen content in water estimated?

(10+10)

- 11. a) Briefly explain the following
 - (i) Love canal episode

(ii) Itai Itai disease

(8)

b) Explain the toxicity of DDT and cadmium.

(6) (4)

c) Discuss Hydrological cycle.

