STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI-86 (For candidates admitted during the academic year 2015–16 & thereafter)

SUBJECT CODE: 15CH/AC/FB33

B.Sc. DEGREE EXAMINATION, NOVEMBER 2017 BRANCH V(a) – PLANT BIOLOGY & PLANT BIOTECHNOLOGY BRANCH VI(a) - ADVANCED ZOOLOGY & BIOTECHNOLOGY THIRD SEMESTER

COURSE PAPER TIME		: ALLIED C : FUNDAMI : 3 HOURS	MAX.MARKS :100 (30x1=30)			
I.C	CHOOSE T	HE CORREC	ANSWER ALL QU CT ANSWER:			
1.	SDS PAGI	E is used to s	eparate macromolecule	es		
	a) Carbohy	drates	b) Lipids	c) Proteins	d) Nucleic acids	
2.	Phenolphthalein turns in acidic solutions					
	a) Pink		b) Red	c) Colorless	d)White	
3.	Haemophilia is a disease					
	a) X linked dominant			b) X linked recessive	e	
	c) Y linked dominant			d) Y linked recessive	e	
4.	PMNL cells is also called as					
	a) Neutrop	hil	b) Eosinophil	c) Lymphocytes	d) Basophil	
5.	pH of bloo	d is				
	a) 6.80-7.2	.1	b)7.35-7.45	c)7.42-7.80	d) 7.26 - 7.84	
6.	The SI unit of entropy is					
	a) Joule		b) Kelvin	c) Kelvin/joule	d) Joule/kelvin	
7.	Starch is hydrolyzed in saliva by					
				c)Maltase	d) Lactase	
8.	is a heteropolysaccharide					
	a) Starch		b) Sucrose	c) Glycogen	d) Hyaluronic acid	
9.	Pick the odd one out of the following					
	a) NAD		b) TPP	c) PPi	d) FAD	
10.		in example of				
	a) Substrat	e	b) Group	c) Relative	d) Reaction	
II.	FILL IN T	HE BLANKS	S:			
11.	pH=	•				
12.	sol	lutions are tho	se with a pH above 7.			
13.	13. Sodium, potassium and bicarbonates are present in the body.					
14.	4 cells involved in immunity.					
15.	5. SI unit of enthalpy is					
16.	5. ATP and ADP are compounds.					
17.	7 is pathway where the glucose is synthesized from non carbohydrate sources.					
18.	3. Starch is made up of amylose and					
19.	9. Optimum temperature is the temperature at which enzyme showsactivity.					
20.	20proposed indued fit model of enzyme action.					

III.STATE WHETHER TRUE OR FALSE:

- 21. The centrifuge works under the sedimentation principle.
- 22. In sickle cell anemia red blood cells carry adequate oxygen throughout the body.
- 23. Endergonic reaction are reactions where energy is released
- 24. Sucrose is made up of glucose and fructose
- 25. Coenzyme is anon protein part of the enzyme.

IV. ANSWER IN ONE OR TWO SENTENCES:

- 26. Ionic product of water
- 27. Acidosis
- 28. Steady state principle
- 29. Chondroitin sulphate
- 30. Active site

SECTION - B

(5x6=30)

Answer any FIVE questions:

- 31. Explain the properties of water.
- 32. Discuss the role of buffers and electrolytes in the body.
- 33. Describe the role of ATP as the energy currency of the cell.
- 34. Explain the oxidation and reduction reactions of glucose.
- 35. Briefly explain the model of enzyme action
- 36. How glucose level is maintained in blood.
- 37. Discuss the digestion and absorption of carbohydrates.

SECTION - C

(2x20=40)

Answer any TWO questions:

- 38. Discuss the steps involved in TCA cycle.
- 39. a) Elaborate on the mechanism of blood coagulation.
 - b) Describe the sequential steps in the electron transport chain of events.
- 40. a) Outline the classification of enzymes with examples?
 - b) Distinguish between
 - i. monosaccharide and disaccharide
 - ii. aldohexose and ketohexose
 - iii.amylose and amylopectin
 - iv. homopolysaccharide and heteropolysaccharide
 - v. Dialysis and sedimentation
