STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI-86 (For candidates admitted during the academic year 2015–16)

SUBJECT CODE: 15CH/AC/FB33

B.Sc. DEGREE EXAMINATION, NOVEMBER 2016 BRANCH V(a) – PLANT BIOLOGY & PLANT BIOTECHNOLOGY BRANCH VI(a) - ADVANCED ZOOLOGY & BIOTECHNOLOGY THIRD SEMESTER

			RE	ZG.NO	•••••			
	RSE : ALLIED (
		ENTALS OF BI	OCHEMISTRY	- I				
TIME	E : 30 MINUT	SECTION – A			MAX.MARKS :30 (30x1=30)			
	AL	L QUESTIONS	TO BE ANSWE	ERED				
		ER ON THE QUI						
I. CI	HOOSE THE CORR	_						
1.	Phenolphthalein is a	colourless, weak	acid which dissoc	ciates in wa	ter forming			
	Phenolphthalein is a colourless, weak acid which dissociates in water forming a) blue anions b) purple anions c) magenta anions d) colourless anions							
	a) oruc amons	purple amons	c) magenta amo	115 (1)	colouriess amons			
2.	The process of sedim	pantation occurs th	prough in respons	e to the for	rae dua to			
۷.								
	a) gravity b) cen	itrifugai accelerati	ion c) electroma	igneusm	d) all of these			
2	7D1 4 1 1'		1 1 1 1					
3.	The metabolic proces		• -					
	a) carbonic acid	b) lactic acid	c) sulphur	ric acid	d) phosphoric acid			
4.		the blood is imp	-		-			
	a) Platelets	b) Hb	c) Plasma		d) Serum			
5.	Compounds in the bi		•	-				
	a) Energy rich compounds		b) High energy compounds					
	c) only a		d) both a & b					
6.	In thermodynamics a	reaction that util	izes energy is cal	led	reaction.			
	a) exergonic	b) endergonic	c) free en	ergy	d) energized			
7.	Though the precursors are produced in the mitochondria, gluconeogenesis takes place							
	in the							
	a) nucleus	b) golgi bodies	c) (cytosol	d) ribosomes			
	,	1,818	,	· J	.,			
8.	The process by which glucose is converted to lactate or pyruvate is called							
	a) uronic pathway	C		1 0				
	a) arome paurway	o) grycory ac pu	un (1) (1) (1)	y cogethy st	s a) gijeogenesis			
9.	Which of the following is a nucleotide coenzyme?							
	a) CoA	b) UDP	c) T	TDD	d) both b & c			
	a) COA	o) ODI	C) I	.11	u, bom b & c			
10.	The energy required for the reactants to undergo a reaction is called							
10.	a) activation energy		_					
	a) activation energy	o) uansinon	C) 1	free energy	d) none of these			

II. FILL UP THE BLANKS:

11.	When water behaves simultaneously like both acid and base the resulting product is					
12.	The pH range of methyl orange is					
13.	Respiratory acidosis is due to the increase in					
14.	The most common sex linked clotting disorder transmitted from females affecting the males is					
15.	The utilizable energy available to do work is called					
16.	Standard free energy is represented by the symbol					
17.	Carbohydrates that completely dissolve in water are called					
18.	The electron transport chain takes place in the					
19.	Substrate concentration at half maximum velocity is called constant.					
20.	The is the non-protein moiety which binds to the apoenzyme.					
III. S	STATE WHETHER TRUE OR FALSE:					
21.	The endogenous waste produced in our body is derived from the oxidation of foodstuff.					
22.	The body can efficiently maintain the acid – base balance when the pH of the blood is 6.4.					
23.	One mole of ATP releases 5.5 kCal of energy.					
24.	Fasting blood glucose in normal individuals is 70-100 mg/dl.					
25.	Tyrosine is a bond specific enzyme.					
IV. A	ANSWER IN ONE OR TWO SENTENCES:					
26.	Define pH.					
27.	What is sickle cell anaemia?					
28.	Differentiate between enthalpy and entropy.					
29.	Draw the Haworths structure for glucose.					
30.	Define apoenzymes.					

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COURSE : ALLIED CORE

PAPER : FUNDAMENTALS OF BIOCHEMISTRY - I

TIME : 2½ HOURS MAX.MARKS : 70

SECTION - B (5x6=30)

Answer any FIVE questions:

- 1. What are the physical properties of water?
- 2. Explain the role of phenolphthalein and methyl orange as indicators.
- 3. Explain the mechanism of coagulation of blood.
- 4. Give the structure of ATP and give reasons why ATP is called the "Energy currency of the cell."
- 5. Elucidate the structure and function of a homopolysaccharide and a heteropolysaccharides.
- 6. Explain the steps in glycogenesis.
- 7. Classify the enzymes based on the reactions they catalyse.

SECTION - C (2x20=40)

(6)

Answer any TWO questions:

- 8. a) Elaborate on the principle, instrumentation, technique and significance of SDS-PAGE and Dialysis. (8+6)
 - b) Describe the maintenance of Glucose level in blood.
- 9. Write the steps involved in the complete breakdown of glucose to CO₂ and H₂O through glycolysis and TCA cycle.
- 10. Write short notes on:
 - a) Lock and Key and Induced Fit Model. (6)
 - b) Maintenance of pH of Blood by Bicarbonate buffer. (6)
 - c) Factors affecting enzyme action. (8)
