STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI 600 086

(For candidates admitted from the academic year 2011-12 & thereafter)

SUBJECT CODE: 11CH/MC/OC64

B.Sc. DEGREE EXAMINATION, APRIL 2015 BRANCH IV - CHEMISTRY SIXTH SEMESTER

					Reg. No	
COURSE PAPER TIME		: MAJOR – CORE				
		: ORGANIC CHEMISTRY - III : 30 MINUTES				
				MAX. MARKS : 30		
			SECTION - A			
			O ON THE QUESTION	PAPER ITSELF.		
ANSW			E QUESTIONS.		(30x1=30)	
Ι	Cł	oose the c	orrect answer.			
1.	Do	action boty	waan NaOU and	viold diablor	acarbana	
1.	3)	CH Cl	ween $NaOH$ and b) CH_2Cl_2	yieid dicilion	d) CCI	
	a)	C113C1	$0)$ $GH_2G\iota_2$	c) ongoi	u) cc14	
2.	Wh	ich of the				
	۵)	CH +	b) $CH_3CH_2^+$	a) (a) +	d) $\left\langle \begin{array}{c} +\\ -\\ -\\ \end{array} \right\rangle$	
	a)	C11 ₃	0) GH ₃ GH ₂	c) $\left\langle \bigcirc \right\rangle$ $-CH_2$	u) \	
3.			can exhibit zwitter ionic.			
	a)	CH_3COOH	H b) CH_3NH_2	$H_3C - CH - COOH$	$H d) H_3C - CH - NH_2$	
				NH_2	NH_2	
4.	Pre	dict the pro	oduct	$\stackrel{\Delta}{\longrightarrow}$?		
	a)		b) [c)	d)	
		·	~	~)	
5.	An	active met	hylene carbon present in			
				0		
	a)	H_3CCH_2	CO CH ₂ COOEt	b) H_3C-C	au.	
				,	CH_2	
				EtOOC		
				Liout		
		CO	00Et			
	c)	H_2C	-	d) all the three		
	,	- \	00Et	,		

6. Ethylene glycol used to protect _____ functional group.



b)
$$C = C$$
 c) $-NH_2$ d) all the three

c)
$$-NH_2$$

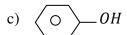
7. β – carotene is a ____

- a) steroid
- b) alkaloid
- d c) terphenoid
- d) drug

8. Acctic anhydride can react with _____ functional group in an alkaloid.

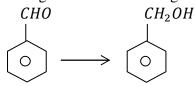
a)
$$R - OH$$

b)
$$R - NH_2$$



d) all the three

9. Choose a reagent for the following conversion



- a) $LiAlH_4$
- b) NaOH
- c) KOH
- d) $KMnO_4$

10. Choose the product for the following reaction

$$H C = C H + H_2C \longrightarrow ?$$

- a) \bigwedge b) $H_3CCH_2CH_3$ c) $H_3C-CH-OH$ d) $H_3C-CH_2-CH_2-CH_3$

II Fill in the blanks.

11. Predict the product:

is ______ acidic than phenol. 12. Picric acid

- 13. Maleic acid and fumaric acid are examples for ______ isomers.
- 14. Ninhydrin can be used to determine _____ acid.
- 15. $RCH_2COCl + RNH_2$ base?
- 16. RCONH₂ \rightarrow RNH₂ is a ______ degradation reaction.
- 17. $H_3C CH_2COCl$? $H_3C CH_2CHO$
- 18. An example for monoterpenoid is _____.
- 19. Total number of Br_2 molecule added to an alkaloid will determine the number of ______.
- 20. Predict the product

$$H_3C - CH = CH - CH_2Cl \xrightarrow{OH^-} A + B$$
? ?

III Match the following:

- 21. Catecol ______ oxidizing reagent.
- 22. Gabriel's synthesis ______ dihydric phenol.
- 23. Schotten Baumann reaction ______ amino acid.
- 24. Cholesterol _____ amide synthesis.
- 25. *KMnO*₄ ______ steroid.

V Answer in a line or two.

26. Predict the product

$$\begin{array}{ccc}
& OsO_4 \\
& OH^-
\end{array}
?$$

- 27. What is Hell-Volhard-Zelinsky reaction?
- 28. State Blane's rule for carboxylic acid.
- 29. Write a method to determine the presence of methoxy group in a natural product.
- 30. Suggest a reagent to protect amino functional group in a chemical.

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COURSE: MAJOR – CORE

PAPER : ORGANIC CHEMISTRY - III

TIME : 2½ HOURS MAX. MARKS : 70

Section - B

Answer any five questions.

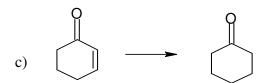
 $(5 \times 6 = 30)$

1. How will you carry out the following conversion?

2. Predict the product

- 3. Write a note on the following with respect to amino acid
 - a) Zwitter ion
- b) Isoelectric point
- 4. a) Explain the mechanism involved in (a) Claisen condensation
 - b) Diel's Alder reaction
- 5. a) How are terpenoids classified?
 - b) State isoprene rule
 - c) What is the basic and common skeleton structure of steroids?
- 6. Write down the synthesis of nicotine.
- 7. How will you carry out the following conversion?

a)
$$H - C \equiv C - H$$
 \longrightarrow $H \subset C = C \subset H$



Section - C

Answer any two questions.

 $(2 \times 20 = 40)$

- 8. a) Explain the concept and emchanism involved in
 - (i) Pinacol-Pinacolone rearrangement
- (ii) Benzilic rearrangement
- b) Write an appropriate reason for the following statement
 - (i) p-nitrophenol more acidic than m-nitrophenol
 - (ii) Propanoic acid is less acidic than mono chloroacetic acid.

- 9. a) Write down the $A_{AC}2$ and $B_{AC}2$ mechanism during hydrolysis of carboxylic ester.
 - b) Using acetoacetic ester synthesise the following compounds.

(i)
$$H_2C-COOH$$
 (ii) $H_3C-C-CH_2CH_3$ $H_2C-COOH$

- 10. a) Elucidate the structure of piperine
 - b) Write down the synthesis of citral.



