STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI-86 (For candidates admitted during the academic year 2008 – 09 & thereafter)

SUBJECT CODE: CH/MC/OC54

B.Sc. DEGREE EXAMINATION, NOVEMBER 2011 BRANCH IV- CHEMISTRY FIFTH SEMESTER

COURSE : MAJOR CORE

PAPER : ORGANIC CHEMISTRY-II

TIME : 30 MINUTES MAX.MARKS: 30

> SECTION - A (30x1=30)

ANSWER ON THE QUESTION PAPER ITSELF.

Answer all the questions.

Ι **Choose the correct answer:**

- 1. Nitration of nitrobenzene with HNO₃/H₂SO₄ gives.
 - (a) m-dinitrobenzene
- (b) o-dinitrobenzene
- (c) p-dinitrobenzene
- (d) Aniline

- 2. Aniline reacts with Bromine in acetic acid to yield.
 - (a) 2,4,6-tribromo aniline

(b) p-bromo aniline

(c) m-bromo aniline

- (d) o-bromo aniline
- 3. Which of the following heterocyclic compound is basic?
 - (a) Furan
- (b) thiophene
- (c) pyrrole
- (d) Pyridine

- 4. An example for azo dye is
 - (a) Phenolphthalein

- (b) methyl orange (c) Crystal violet (d) Malachite green.
- 5. Which of the following is not an antibiotic
 - (a) Penicillin
- (b) streptomycin
- (c) Chloromycetin
- (d) Aspirin

- 6. Ouinine can be used as
 - (a) Analgesic
- (b) Antimalarials
- (c) Hypnotics
- (d) Sedatives

II State whether the following statements are true or false.

- 7. Aniline reacts with NaNO₂/HCl at 0 °C to yield Benzene diazonium chloride.
- 8. The Magnesium is the central metal atom in hemoglobin.
- 9. Malachite green is a triphenylmethane dye.
- 10. In acidic medium phenolphthalein is pink in colour.
- 11. Para Amino benzoic Acid is structurally similar to sulpha drugs..
- 12. Tetracycline is an antibiotic drug.

III Match the following:

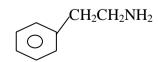
- 13. Hinsberg method Natural Dye 14. Skraup's synthesis - Aniline
- 15. Alizarin Separation of amines
- 16. Reduction of nitrobenzene Quinoline 17. Acetyl salicylic acid - Azo dye 18. Methyl orange - Aspirin

IV Fill in the blanks:

- 19. The isomeric form of CH₃NO₂ is ______?
- 20. $RNH_2 + RCH_2COC1 \xrightarrow{base} ?$
- 21. $\underbrace{\hspace{1cm}}_{O} \xrightarrow{acetylnitrate} \xrightarrow{?}$.
- 22. The structure of indole is ______.
- 23. An example for a dye which are used as an acid-base indicator is _____.
- 24. Lysergic acid diethyl amide (LSD) can be used as ______ drugs.

V Answer in a line or two:

- 25. Why is triethylamine less basic than diethylamine?
- 26. Predict the products formed for the following reaction. $CH_3CH_2NH_2 + HONO \longrightarrow ?$
- 27. What is Indigo?
- 28. Write down the structure of Alizarin.
- 29. What do you understand by the term antipyretics?
- 30. How will you synthesise isoquinoline from



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B.Sc. DEGREE EXAMINATION, NOVEMBER 2011

BRANCH IV- CHEMISTRY FIFTH SEMESTER

COURSE : MAJOR CORE

PAPER : ORGANIC CHEMISTRY-II

TIME : 2½ Hours MAX.MARKS : 70

SECTION – B (5x6=30) ANSWER ANY FIVE QUESTIONS

- 1. How will you synthesise m-nitro aniline from benzene?
- 2. How does aniline react with the following reagents?
 - (a) Acetic anhydride (b) Nitrous acid at O-5°C (c) Benzene diazonium chloride
- 3. Why furan, thiophene and pyrrole are aromatic in nature? Explain.
- 4. Among pyridine and pyrrole, which is more basic? Explain.
- 5. How will you classify dyes based on chemical structure?
- 6. Write down the synthesis of sulphapyridine and sulphadiazine
- 7. Write down the therapeutic applications for the following pharmaceutical drugs.
 - (a) aspirin
- (b) sulphaguanidine

(c) penicillin

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SECTION – C ANSWER ANY TWO QUESTIONS

- 8. (a) How will you distinguish primary, secondary and tertiary amines?
 - (b) How will you separate a mixture of amines?

(10+10)

(2x20=40)

- 9. (a) Elucidate the structure of quinoline and write down its synthesis.
 - (b) Write a note on the structure of Chlorophyll.

(10+10)

- 10. (a) Write a preparation of the following dyes.
 - (i) Malachite green.
- (ii) Methyl orange.
- (b) Write a note on the application of dyes in leather, cosmetics, textile and food industries. (12+8)
- 11. (a) Explain the mode of action of sulphadrugs.
 - (b) Write a note on Hypnotics, Sedatives and Psychedelic drugs.
 - (c) How will you convert Aniline to s-Tribromobenzene. (6+9+5)