STELLA MARIS COLLEGE (AUTONOMOUS), CHENNAI 600 086

B.Sc. DEGREE : BRANCH IV-CHEMISTRY

COURSE SCHEDULE

SEMESTER I

Subject Code	Title of Course
23CH/MC/GC14	General Chemistry
23CH/MC/P112	VOLUMETRIC ANALYSIS PRACTICAL
23CH/SS/HC13	LIFE SKILLS- HEALTH, ENERGY AND COMPUTER BASICS

STELLA MARIS COLLEGE (AUTONOMOUS), CHENNAI COURSE PLAN

June - November 2024

Department : Chemistry

Name/s of the Faculty : Dr.Mary George
Course Title : General Chemistry
Course Code : 23CH/MC/GC14

Shift : I

COURSE OUTCOMES (COs)

COs	Description	CL
CO1	Recollect the fundamentals of periodicity of elements, atomic structure, acids and bases, nuclear chemistry and organic chemistry	K1, K2
CO2	Calculate eigen values and eigen functions, categorize periodic properties of s, p and d elements, compare radioactive disintegration series and differentiate the types of reactive intermediates in organic chemistry	К3
CO3	Apply the concepts of periodicity of elements, dual nature of light, electromagnetic spectrum, quantum numbers, acid-base theories, half-life of disintegration, nomenclature and classification of organic compounds to solve problems	K4
CO4	Analyze elements based on their atomic structure and periodic properties, classify radioactive elements based on their stability, binding energy and mass defect and develop mechanisms based on organic reagents and reactions.	K5
CO5	Evaluate the effect of electronic displacements in covalent bonds on molecular stability and chemical reactivity, investigate the applications of nuclear chemistry, explain the chemical behavior of elements from their periodic properties and solve problems in atomic structure	К6

Week	Unit No.	Content	Cognitive Level	Teaching Hours	COs	Teaching Learning Methodology	Assessment Methods
Jun 24 – 26, 2024 (Day Order 4 - 6)	1.1	Rutherford's nuclear model of the atom. Planck's quantum theory of radiation. Photoelectric effect. Bohr's theory, its limitations and atomic spectrum of hydrogenatom. Wave mechanics: de Broglie equation, Davisson - Germer experiment. Heisenberg's principle of uncertainty. ComptonEffect.	K1-4	5	CO1-4	lecture and discussion	Short questions
Jun 27 – July 4, 2024 (Day Order 1 - 6)	1.3	Postulates of Quantum mechanics; operators- Hermitian operators, Laplacian and Hamiltonian operators, Eigen functions and Eigen values of operators. Conditions for a well- behaved function, Schrodinger wave equation (no derivation). Significance of ψ and ψ^2	K3,K4	5	CO3	lecture and discussion	Problem solving

July 5 – 12, 2024 (Day Order 1 - 6)	2.1	Quantum numbers and their significance. Normalized and orthogonal wave functions. Sign of wave functions. Shapes of orbitals Periodic Table, horizontal, vertical and diagonal relationships in the periodic table — Li-Mg, Be-Al, B-Si	K4	5	CO3	Lecture and Discussion	Worksheet-Quiz
July 15 – 23, 2024 (Day Order 1 - 6)	2.2	Periodicity of properties of s, p and d – block elements with respect to atomic radii, ionic radii, covalent radii, ionization energy Periodicity of properties of s, p and d - block elements with respect to electronegativity, electron affinity	K1-K4, K6	5	CO1-4	Power point presentation	Short answer questions- Worksheet
July 24 – 31, 2024 (Day Order 1 - 6)	3.1	Inert pair effect, effective nuclear charge – screening effect, Slater rules Concepts: Bronsted-Lowry, Lux-Flood, Solvent -system and Lewis Relative strength of acids and bases: Effect of solvent, levelling effect, polarity and substituents	K1-K4	5	CO1-4	Lecture &Discussion	Component MCQ TEST (20 marks) UNIT 1 and UNIT 2 Worksheet

Aug 1 – 5, 2024 (Day Order 1 - 3)	3.2	HSAB - Principle and its applications	K1-K4	2	CO1-4	Demonstration experiment	Worksheet
Aug 6 – 10, 2024				C.A. Tes	st – I Unit 1	,2&3	
Aug 12 – 14, 2024 (Day Order 4-6)	4.1	Elementary Particles - Concept of Nuclides, representation of isobars, isotones, isotones with examples. Nucleus structure – Liquid Drop and Shell Model. Nuclear stability – <i>n/p</i> ratio, binding energy, mass defect and magic numbers	K1-K4	3	CO1-4	Lecture & Discussion	Worksheet
Aug 16 – 23, 2024 (Day Order 1-6)	4.2	Radioactive elements, modes of decay – Neutron, Positron Theory of α , β and γ emission, characteristics of α , β and γ particles, K-electron capture and positron emission. Half-life period, Geiger – Nuttall rule	K1-K4	5	CO1-4	Lecture &Discussion	Short answer
Aug 27 – Sep 3, 2024 (Day Order 1-6)	4.2	Radioactive displacement laws – Soddy, Fajan and Russel. Radioactive decay series 4n, 4n+1, 4n+2 and 4n+3	K4	5	CO2-3	Lecture &Discussion Field Visit	Short answers

Sep 4 – 11, 2024 (Day Order 1-6)	4.3	Detection and measurement of radioactivity - Ionization chamber, Geiger- Muller counter and Scintillation counter. Artificial radioactivity - Artificial transmutation of elements, nuclear reactions - nuclear fusion and fission	K1-K5	5	CO4-6	Discussion Field Visit	Component Presentation/ Assignment (10 marks) Unit 4.2 &4.3
Sep 12 - 20, 2024 (Day Order 1-6)	5.1	IUPAC nomenclature of organic compounds Types of organic reaction and reagents: Nature of bond fission – Homolytic and Heterolytic	K1-K4	5	CO1-3	Lecture &Discussion	Problem solving- Worksheet
Sep 23 - 26, 2024 (Day Order 1-4)	5.2	Types of reagents – Electrophiles and Nucleophiles. Substitution, Addition, Elimination and Rearrangement reactions (definition with anexample) Reactive intermediates with examples – Carbocations, Carbanions and Free Radicals – Conditions favouring their formation, stability and structure, their reactions with example	K1-K5	3	CO1-5	Lecture &Discussion	Component MCQ TEST (20 marks) UNIT 4 and UNIT 5.1
Sep 27 – Oct 3, 2024				C.A. Test -	- II Unit 48	£5.1-5.3	

Oct 4 – 5, 2024 (Day 5 & 6)	5.3	Electron displacement effects - Inductive, Electromeric, Mesomeric, Resonance, Hyperconjugation Steric effects, Tautomerism Concept of Aromaticity – Definition, Hückel's Rule	K4-K6	2	CO1-3	Lecture &Discussion	Worksheet
Oct 7 - 15, 2024 (Day Order 1 to 6)	5.4	Application to benzenoid and non-benzenoid compounds Benzene, naphthalene,	K4-K6	5	CO3-5	Lecture &Discussion	Worksheet
Oct 16 - 22, 2024 (Day Order 1 to 6)	5.4	cyclopropenyl cation, cyclopentadienyl anion and tropyliumcation	K4-K6	5	CO3-5	Lecture &Discussion	Worksheet
Oct 23 - 24, 2024 (Day Order 1 to 2)				RI	EVISION		

STELLA MARIS COLLEGE (AUTONOMOUS), CHENNAI

COURSE PLAN June - November 2024

Department : CHEMISTRY

Name/s of the Faculty : DR. MARY TERESITA V* & DR. JANET SABINA X

Course Title : VOLUMETRIC ANALYSIS PRACTICAL

Course Code : 23CH/MC/P112

Shift : I

COURSE OUTCOMES (COs)

COs	Description	CL
CO1	Recollect the principles of different titrations and calibration, concepts of molarity, normality and equivalent weight	K1, K2
CO2	Differentiate between acids/bases and oxidising/reducing agents	K2
CO3	Calculate the molarity, normality and equivalent weights of acids, bases, oxidising and reducing agents	К3
CO4	Categorize indicators based on the type of titration and pH	K4
CO5	Estimate the amount of a metal ion/acid/base present in the whole of the given solution	K5,K6

Week	Unit No.	Content	Cognitive Level	Teaching Hours	COs	Teaching Learning Methodology	Assessment Methods
Jun 24 – 26, 2024 (Day Order 4 - 6)		No Class					
Jun 27 – July 4, 2024 (Day Order 1 - 6)	2	Volumetric Analysis Theory and Principle behind the different titrations, equivalent weight calculations, concentration terms-normality, molarity and molality	K1-K4	3	1-4	Lecture and Discussion	Quiz/Test/Group Discussion
July 5 – 12, 2024 (Day Order 1 - 6)	1	Calibration of Burettes / Pipettes	K1-K2	3	1-2	Demonstratio n & Hands on training	-
July 15 – 23, 2024 (Day Order 1 - 6)	3	Estimation of Na ₂ CO ₃ / HCl	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
July 24 – 31, 2024 (Day Order 1 - 6)	3	Estimation of Na ₂ CO ₃ / HCl	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks

Aug 1 – 5, 2024 (Day Order 1 - 3)	3	Estimation of Oxalic Acid (Permanganimetry)	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
Aug 6 – 10, 2024			C.A. Test	- I	1		
Aug 12 – 14, 2024 (Day Order 4-6)		No Class					
Aug 16 – 23, 2024 (Day Order 1-6)	3	Estimation of Oxalic Acid (Permanganimetry)	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
Aug 27 – Sep 3, 2024 (Day Order 1-6)	3	Estimation of Dichromate (Iodometry)	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks

Sep 4 – 11, 2024 (Day Order 1-6)	3	Estimation of Iron (Dichrometry / Permanganimetry)	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
Sep 12 - 20, 2024 (Day Order 1-6)	3	Estimation of Iron (Dichrometry / Permanganimetry)	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
Sep 23 - 26, 2024 (Day Order 1-4)	3	Estimation of Magnesium / Zinc (Complexometry)	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
Sep 27 – Oct 3, 2024				C.A. 7	Γest - II	_ 	
Oct 4 – 5, 2024 (Day 5 & 6)		No Class					

Oct 7 - 15, 2024 (Day Order 1 to 6)	3	Estimation of Magnesium / Zinc (Complexometry)	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
Oct 16 - 22, 2024 (Day Order 1 to 6)	3	Estimation of Chloride (Argentometry), Estimation of Hardness	K1-K6	3	1-6	Hands on experiment	Principle & Procedure = 10 marks Experiment (40 marks) Up to 2% error - 40 marks 2.1 - 3.0% error - 35 marks 3.1 - 4.0% error - 25 marks 4.1 - 5% error - 20 marks Above 5% - 15 marks Total = 50 Marks
Oct 23 - 24, 2024 (Day Order 1 to 2)				REV	ISION	1	

STELLA MARIS COLLEGE (AUTONOMOUS), CHENNAI

Course Schedule: June - November 2024

Department : CHEMISTRY

Name/s of the Faculty : Dr. REVATHY RAJAGOPAL & Dr. Avila Josephine B

Course Title : LIFE SKILLS- HEALTH, ENERGY AND COMPUTER BASICS

Course Code : 23CH/SS/HC13

Shift : I

eek No. of	Units & Topics	Teaching Methodology	Text & References	Method of Evaluation

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