STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI 600 086 (For candidates admitted during the academic year 2016 – 2017 & thereafter)

B.Voc. DEGREE EXAMINATION, APRIL 2024 FOOD PROCESSING AND QUALITY CONTROL FOURTH SEMESTER

COURSE : MAJOR CORE

PAPER : ANALYTICAL TECHNIQUES IN FOOD QUALITY ASSURANCE

SUBJECT CODE: 16VF/VM/AT46

TIME : 5 HOURS MAX.MARKS: 100

(Theory: 50marks +Practical: 50 marks)

	SECT	ION – A	
	VER ALL QUESTIONS:		$(20 \times 1 = 20)$
	oose the Correct answer:		
1.	In thin layer chromatography, the stationary	phase is	_ :
	a) solid b) liquid	c) ion	d) gas
2.	 a) solid b) liquid Hardness is due to the presence of	salts.	
	a) Calcium and Magnesium ions	c) Tungsten and Zirco	onium ions
_	b) Sodium and Potassium ions	d) None of the above	
3.	Abbe's refractometer is a type	of refractometer.	
	a) Benchtop b) traditional		
4. Which of the following is not used as a detector in Flame emission photometers?			
	a) Photovoltaic cellsb) Photoemissive cells	c) Chromatogram	
_	,	, I	
5.	What is the wavelength range corresponding		
	a) 400 – 800 nm b) 200 – 400 nm	c) 200 – 800 nm	d) 400 – 800 nm
 II Fill in the blanks: 6. Turbidity is measured using a 7. Functional groups responsible for light absorption are called 8. The chlorides and sulphates of calcium and magnesium cause 			
9.	Temperatures vs is measured in differential scanning calorimetry.		
10. Total dissolved solid content can be reduced by			
111 Sta 11 12 13 14	ate whether True or False: Thermally stable samples can be analysed b. Beer Lambert's law can be used for both dil. Column chromatography is a type of liquid. Viscometry can be used to determine the pH. The Karl-Fischer titration is used to determine	y Gas chromatography ute and concentrated so liquid chromatography I of a water sample.	olutions. technique.
16	swer in a sentence: . What is Chemical Oxygen Demand? . Give one advantage of High Performance Li	iguid Chromatography	

- 17. Give one advantage of High Performance Liquid Chromatography.
- 18. Define retention factor.
- 19. What is the mobile phase in Gas chromatography?
- 20. What is the Becke line method?

SECTION B

Answer any SIX questions:

 $(6 \times 3 = 18)$

- 21. Explain the principle of photocolorimetry.
- 22. Differentiate between stationary and mobile phase.
- 23. Discuss the principle of Fluorimeter.
- 24. Define Viscometry and Refractive index.
- 25. Give any two applications of chromatography.
- 26. Define hardness. Differentiate between permanent and temporary hardness.
- 27. What is water quality monitoring? What is its impact on food safety?
- 28. Define elution and retention time.
- 29. What is the principle of paper chromatography?
- 30. State Beer-Lambert's law. Give the equation for the same.

SECTION C

Answer any TWO questions:

 $(2 \times 6 = 12)$

- 31. Explain the principle and instrumentation of Flame photometry. How is sodium estimated using Flame photometer?
- 32. Discuss the instrumentation of a UV-visible double beam spectrophotometer with a neat diagram.
- 33. Explain the principle and working of high performance liquid chromatography.
- 34. Discuss the importance of water in food processing.
