

B.Sc. DEGREE EXAMINATION, NOVEMBER 2021

THIRD SEMESTER

COURSE: ALLIED CORE

PAPER: FUNDAMENTALS OF CHEMISTRY-I

MAX .MARKS: 100

TIME: 3 HOURS

SECTION-A

Answer all the questions

(15 x 2 =30 Marks)

I. Match the following:

- | | |
|---------------------|-------------------------|
| 1. Cis-polyisoprene | a. Disaccharide |
| 2. Tyrosine | b. Lewis base |
| 3. Maltose | c. Condensation polymer |
| 4. Ammonia | d. Aromatic amino acid |
| 5. Bakelite | f. Natural rubber |

II. Fill in the blanks:

- Expansion of PTFE is _____
- Structure of a carbocation is _____
- The characteristic functional group present in compounds that are used as food preservatives is _____
- Decomposition of acetaldehyde is a _____ order reaction.
- pH of human blood is maintained by _____ buffer system.

III. State true or false:

- Unit of zero order rate constant is $\text{mol L}^{-1}\text{sec}^{-1}$
- Heterolytic fission results in the formation of free radicals
- Oxalic acid is a polyprotic acid.
- Radioactive disintegration follows first order kinetics.
- Cysteine is an aliphatic amino acid.

SECTION-B

IV. Answer any five :

(5x8=40 Marks)

- Discuss the free radical mechanism of addition polymerisation.
- State Ostwald's Dilution law. Derive the expression for K_a and highlight the significance of its relation to α .
- a) Differentiate between thermoplastic and thermosetting plastics using suitable examples.
b) Explain the process of vulcanisation of rubber. (4+4)

