

STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI – 600 086.  
(For candidates admitted during the academic year 2008-09 & thereafter)

SUBJECT CODE : PH/MC/PA14

B.Sc. DEGREE EXAMINATION NOVEMBER 2009  
BRANCH III - PHYSICS  
FIRST SEMESTER

REG. No. \_\_\_\_\_

COURSE : MAJOR – CORE  
PAPER : PROPERTIES OF MATTER AND ATOMIC PHYSICS  
TIME : 30 MINS. MAX. MARKS : 30

SECTION – A

TO BE ANSWERED IN THE QUESTION PAPER ITSELF

ANSWER ALL QUESTIONS: ( 30 x 1 = 30 )

I CHOOSE THE CORRECT ANSWERS:

- Hooke's law is within elastic limit  
a) Stress / strain = E  
b) Strain/ Stress = E  
c) Rigidity modulus/bulk modulus=E  
d) Force/Area=E
- What is the dimension of strain?  
a)  $ML^{-1}T^{-2}$   
b)  $ML^{-2}T^{-2}$   
c)  $MLT^{-2}$   
d) NIL
- The ratio of lateral strain to longitudinal strain is  
a) Young's modulus  
b) Bulk modulus  
c) Poisson's ratio  
d) Rigidity modulus
- What is the unit of coefficient of viscosity?  
a)  $N/m^2$   
b)  $N/m^2/unit\ vel.\ gradient$   
c)  $N/m$   
d)  $Kg/s$
- Excess pressure inside a liquid drop is  
a)  $T/r$   
b)  $2T/r$   
c)  $3T/r$   
d)  $4T/r$
- Angle of contact of glass with mercury is  
a)  $140^\circ$   
b)  $0^\circ$   
c)  $90^\circ$   
d)  $45^\circ$
- Adhesive force is between the  
a) same molecules  
b) different molecules  
c) same charges  
d) different charges
- The dimensional formula for surface tension is  
a)  $MLT^{-2}$   
b)  $MT^{-2}$   
c)  $M^{-1}L^{-1}T^{-1}$   
d)  $ML^{-1}T^{-1}$
- The flow of liquid above and below the critical velocity is  
a) streamline flow  
b) turbulent flow  
c) streamline & turbulent flow  
d) turbulent & streamline flow

10. Artificial rain is sometime brought about in the saturated cloud by injecting solid  
 a) CO<sub>2</sub>                      b) SO<sub>2</sub>                      c) NaCl                      d) Na<sub>2</sub>CO<sub>3</sub>
11. Canal rays are also called  
 a) α - rays                      b) β - rays                      c) γ - rays                      d) positive rays
12. Einstein's photo electric equation is  
 a)  $h\nu = h\nu_0$                       b)  $h\nu = h\nu_0 + \frac{1}{2} m v^2$   
 c)  $\frac{1}{2} m v^2 = h\nu$                       d)  $\frac{1}{2} m v^2 = h\nu_0$
13. The fundamental postulates of vector atom model is  
 a) spatial quantization  
 b) spinning of electron  
 c) spatial quantization and spinning of electron  
 d) the angular momentum of quantized orbit is  $nh/2\pi$
14. Normal energy state of the Hydrogen atom is  
 a)  $13.6/n^2$  eV                      b)  $-13.6/n^2$  eV  
 c)  $n^2/13.6$  eV                      d)  $-n^2/13.6$  eV
15. Compton's change in wave length is  
 a)  $d\lambda = h^2 (1 - \cos \theta) / m_0 c$                       b)  $d\lambda = h^2 (1 - \cos \theta) / m_0 c^2$   
 c)  $d\lambda = h^2 (1 - \cos \theta) / m_0^2 c^2$                       d)  $d\lambda = h (1 - \cos \theta) / m_0 c$

## II FILL IN THE BLANKS:

16. Mass number is \_\_\_\_\_
17. Moseley's law is \_\_\_\_\_
18. Bragg's law is \_\_\_\_\_
19. Increase in temperature will \_\_\_\_\_ the S.T. of liquid.
20. Splitting of spectral lines under the magnetic field is \_\_\_\_\_

## III STATE WHETHER TRUE OR FALSE:

21. The angular momentum of Bohr's quantized orbit is  $nh/2\pi$ .
22. In a highly viscous liquid, the velocity of a body is non uniform after it attains terminal velocity.
23. No two electrons in an atom exists in the same quantum state.
24. The Newton's viscous flow law is  $F = -\eta A dv/dt$ .
25. The Reynold's no. determines the nature of flow whether turbulent or steam line flow of a Liquid through a narrow tube.

**IV ANSWER BRIEFLY:**

26. What is bending moment?

27. What is Compton effect ?

28. What is Bohr Magneton ?

29. Why rain drops or dew drops assume spherical shape ?

30. What is Stark effect?

XXXXXXXX

STELLA MARIS COLLEGE (AUTONOMOUS) CHENNAI – 600 086.  
(For candidates admitted during the academic year 2008-09 & thereafter)

SUBJECT CODE : PH/MC/PA14

**B.Sc. DEGREE EXAMINATION NOVEMBER 2009**  
**BRANCH III - PHYSICS**  
**FIRST SEMESTER**

**COURSE : MAJOR – CORE**  
**PAPER : PROPERTIES OF MATTER AND ATOMIC PHYSICS**  
**TIME : 2 ½ HOURS** **MAX. MARKS : 70**

**SECTION – B**

**ANSWER ANY FIVE QUESTIONS: ( 5 x 5 = 25 )**

1. Calculate the value of  $\sigma$  for the material, given Young's Modulus,  
 $q = 12.25 \times 10^{10} \text{ N / m}^2$ . Rigidity modulus,  $n = 4.55 \times 10^{10} \text{ N / m}^2$ .
2. Calculate the work done in stretching a uniform metal wire of area of cross section  $10^{-6} \text{ m}^2$  and length 1.5 m through  $4 \times 10^{-3} \text{ m}$ . Given  $q = 2 \times 10^{11} \text{ N / m}^2$ .
3. Calculate the work done in spraying spherical drop of water of  $10^{-3} \text{ m}$  radius into million droplets all of the of the same size, the S.T. of water is  $72 \times 10^{-3} \text{ N / m}$ .
4. Water flows through a horizontal tube of length 0.2 m and internal radius  $8.1 \times 10^{-4} \text{ m}$ . Under a constant head of the liquid 0.2 m height. In 12 minutes  $8.64 \times 10^{-4} \text{ m}^3$  of liquid comes from the tube. Calculate the coefficient of viscosity of water. Density of water =  $1000 \text{ kg / m}^3$  and  $g = 9.8 \text{ m/s}^2$ .
5. The spacing between principal planes of NaCl crystal is 2.82Å. It is found that first order Bragg reflection occurs at an angle of  $10^\circ$ . What is the wave length of X- rays?
6. State the laws of photo electric effect.
7. Explain excitation and ionization potential with an example.

**SECTION – C**

**ANSWER ANY THREE QUESTIONS: ( 3 x 15 = 45 )**

8. Define Rigidity modulus. Describe with necessary theory, how would you determine the rigidity modulus of a wire experimentally by using Torsional pendulum.
9. Define surface tension. Describe Jaegar's method of studying surface tension of water with temperature.

10. Define the coefficient of viscosity. Derive Poiseuille's formula for the rate of flow of liquid through a capillary tube. Describe laboratory method for determining coefficient of viscosity of a liquid at room temperature.
11. What is mass spectrograph? Describe the construction, working and theory of a Dempster mass spectrograph.
12. What is Zeeman Effect? Describe the experimental arrangement for studying the Zeeman Effect.

XXXXXXXXXX